

Satd aq NaCl (50 ml) was added, the soln was extd with  $C_6H_6$  ( $4 \times 100$  ml), the solvent was evapd, and the tacky residue was chromatographed (Florisil, petr ether, bp 60–100°). The viscous product weighed 7.54 g (80% yield). It was dissolved in EtOH (100 ml) and HCl (7.7 ml) and hydrogenated at 3.71 kg/cm<sup>2</sup> in the presence of 5% Rh/C (3 g) at 45° for 45 hr. Filtration from the catalyst, evapn of EtOH, addition of LiOH, extn with  $CHCl_3$ , and evapn of the solvent yielded oily 1, di(5-nitrobarbiturate): yellow crystals from  $H_2O$  (5.2 g, 62%); mp 251–252°. *Anal.* ( $C_{26}H_{28}N_2O_{11} \cdot H_2O$ ) C, H, N. The di-*dl*-tartrate was prepd. in  $Me_2CO$  and crystd from MeOH, mp 161–162°. *Anal.* ( $C_{26}H_{34}N_2O_6$ ) C, H, N. Mass spectrum of liberated oily 1 ( $C_{12}H_{12}N_2$ ) revealed *m/e* 194.

**3-(2-Pyridyl)-3-quinuclidinol (3b).**—A soln of 55 g (0.35 mole) of 2-bromopyridine in 50 ml of  $Et_2O$  was added rapidly to a stirred soln of *n*-BuLi (0.35 mole) in hexane at –50° under  $N_2$ . After stirring for 20 min, a soln of 3-quinuclidinone (12.5 g, 0.1 mole) in  $Et_2O$  (100 ml) was added dropwise. The mixt was stirred for 2.5 hr while the temp was allowed to rise gradually to 25°, poured on ice-AcOH, and extd with  $Et_2O$ . The  $H_2O$  layer was made ammoniacal and extd exhaustively with  $CHCl_3$ . The exts were dried ( $Na_2SO_4$ ) and evapd, and the residue was crystd from  $Me_2CO$ ; yield 8.5 g (42%); mp 163–164°. The product sublimed at 130° (0.1 mm). *Anal.* ( $C_{12}H_{16}N_2O$ ) C, H, *m/e* 204.

**3-(2-Pyridyl)-1-azabicyclo[2.2.2]-2-octene.**—An intimate mixt of **3b** (4.7 g, 23 mmoles) and powdered potassium pyrosulfate (40 g) was fused at 240° for 1 min.<sup>7</sup> After cooling, the melt was treated with ice, and the soln was made ammoniacal and extd exhaustively ( $CHCl_3$ ). Evapn of the solvent and chromatography using Florisil and hexane yielded 2.87 g (64%) of colorless crystals, mp 75–76°. *Anal.* ( $C_{12}H_{14}N_2$ ) C, H, *m/e* 186.

**3-(2-Piperidinyl)quinuclidine (2).**—A soln of 4.5 g (24 mmoles) of the unsatd pyridine deriv in EtOH (100 ml) and HCl (4.5 ml) was hydrogenated at 25° and 3.51 kg/cm<sup>2</sup> with  $PtO_2$  (0.58 g) for 1.5 hr. The soln was filtered and evapd, and the residue was made basic with LiOH and extd with  $CHCl_3$ . The oily residue from the ext was treated with 600 ml of an EtOH soln of 5-nitrobarbituric acid (8.65 g, 0.05 mole). A ppt which formed immediately was collected, dried, and recrystd from  $H_2O$  (200 ml). Yellow rosettes (3.54 g) had mp 252–254° dec. The base was regenerated (aq KOH), extd ( $Et_2O$ ), and crystd (anhyd  $Et_2O$ ) at –50°; yield 1.1 g; mp 61–62° (diastereomer **2A**); tlc [ $Al_2O_3$ ,  $CHCl_3$ - $Et_2NH$  (5%)]  $R_f$  0.4. *Anal.* ( $C_{12}H_{22}N_2$ ) C, H, N; *m/e* 194.

**Diastereomer 2B.**—The aq mother liquor of the diluturate was coucd to 50 ml and cooled to 0°. It deposited 4.86 g of yellow needles, mp 261–262° dec. The diastereoisomeric base was regenerated and recrystd as above; yield 1.7 g; mp 69–70°; tlc as above,  $R_f$  0.6. *Anal.* ( $C_{12}H_{22}N_2$ ) C, H, N; *m/e* 194.

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(7) Cf. F. J. Villani and C. A. Ellis, *J. Med. Chem.*, **13**, 1245 (1970).

## Deamino-D-oxytocin<sup>1</sup>

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We have previously reported<sup>2</sup> that the optical antipode of the posterior pituitary hormone oxytocin did not possess detectable avian vasodepressor or oxytocic

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(2) G. Flouret and V. du Vigneaud, *J. Amer. Chem. Soc.*, **87**, 3775 (1965).

activity, whereas oxytocin possesses approximately 500 units/mg of each of these activities. No indication of an inhibitory effect of D-oxytocin on these activities of oxytocin could be detected. Since deamino-oxytocin is much more potent than oxytocin, it was decided to synthesize deamino-D-oxytocin and test it for these biological activities.

For the synthesis of deamino-D-oxytocin, the desired protected polypeptide amide precursor *S*-Bzl- $\beta$ -mercaptopropionyl-D-Tyr-D-Ile-D-Gln-D-Asn-D-Cys(Bzl)-D-Pro-D-Leu-Gly-NH<sub>2</sub> (I) was prepared by the nitrophenyl ester method,<sup>3</sup> as employed for the synthesis of deamino-L-oxytocin,<sup>4</sup> starting with *Z*-D-Tyr(Bzl)-D-Ile-D-Gln-D-Asn-D-Cys(Bzl)-D-Pro-D-Leu-Gly-NH<sub>2</sub> (II).<sup>2</sup>

The protected polypeptide amide I was treated with Na in liquid  $NH_3$ ,<sup>5</sup> and the resulting disulfhydryl compound was oxidized in dil aq soln with  $K_3Fe(CN)_6$ .<sup>4</sup> After removal of ferro- and ferricyanide ions with the ion-exchange resin AG3-X4 (in the  $Cl^-$  form), the soln gave a negative Ellman test.<sup>6</sup> The deamino-D-oxytocin thus obtd was purified by countercurrent distribution<sup>7</sup> or by partition chromatography<sup>8</sup> on Sephadex G-25 as described in the Experimental Section.

No avian vasodepressor<sup>9</sup> or oxytocic activity<sup>10</sup> was detected upon bioassay of deamino-D-oxytocin, whereas crystalline deamino-L-oxytocin possesses 975 units/mg of avian vasodepressor and 803 units/mg of oxytocic activity.<sup>11</sup> No indication of an inhibitory effect of deamino-D-oxytocin on the above activities of oxytocin could be detected.

## Experimental Section<sup>12</sup>

***S*-Bzl- $\beta$ -mercaptopropionyl-D-Tyr-D-Ile-D-Gln-D-Asn-D-Cys(Bzl)-D-Pro-D-Leu-Gly-NH<sub>2</sub> (I). Procedure A.**—A suspension of 0.6 g of II in 25 ml of  $F_3CCH_2OH$ <sup>13</sup> was saturated with HBr gas previously passed through towers of naphthalene and anhydrous  $CaCl_2$ . After 30 min the resulting soln was evapd to dryness, the residue was redissolved in  $F_3CCH_2OH$ , and the soln was evapd to dryness again. The solid residue obtained was dissolved in 40 ml of MeOH and neutralized with ion-exchange resin IRA-410 (OH). The suspension was filtered, and the filtrate, which gave a negative test for  $Br^-$  ( $AgNO_3$ ), was evapd under vacuum and dried under vacuum over  $P_2O_5$ . A soln of 0.38 g of the free octapeptide thus obtained in 2 ml of DMF was treated with 0.17 g of *p*-nitrophenyl *S*-benzyl- $\beta$ -mercaptopropionate. The waxy material formed overnight was treated with EtOH, filtered off, and washed 3 more times with EtOH; yield

(3) M. Bodanszky, *Nature (London)*, **175**, 685 (1955).

(4) V. du Vigneaud, G. Winestock, V. V. S. Murti, D. B. Hope, and R. D. Kimbrough, Jr., *J. Biol. Chem.*, **235**, PC 64 (1960); D. B. Hope, V. V. S. Murti, and V. du Vigneaud, *ibid.*, **237**, 1563 (1962).

(5) R. H. Sifferd and V. du Vigneaud, *ibid.*, **108**, 753 (1935).

(6) G. L. Ellman, *Arch. Biochem. Biophys.*, **82**, 70 (1959).

(7) L. C. Craig, W. Hausmann, E. H. Ahrens, Jr., and E. J. Harfenist, *Anal. Chem.*, **23**, 1236 (1951).

(8) D. Yamashiro, *Nature (London)*, **201**, 76 (1964).

(9) Avian vasodepressor assays were performed on conscious chickens according to the method of R. A. Munsick, W. H. Sawyer, and H. B. van Dyke, *Endocrinology*, **66**, 860 (1960).

(10) Oxytocic assays were performed according to the method of P. Holton, *Brit. J. Pharmacol.*, **3**, 328 (1948), as modified by R. A. Munsick, *Endocrinology*, **66**, 451 (1960), on isolated uteri from rats in natural estrus with the use of Mg-free van Dyke-Hastings solution.

(11) B. M. Ferrier, D. Jarvis, and V. du Vigneaud, *J. Biol. Chem.*, **240**, 4264 (1965).

(12) All melting points are corrected capillary melting points and were taken on a Thomas-Hoover melting point apparatus. Where analyses are indicated only by symbols of the elements, analytical results obtained for the elements were within  $\pm 0.4\%$  of the theoretical values. All protected peptides were dried to const wt over  $P_2O_5$  under vacuum at 100° before yields were determined.

(13) D. Yamashiro, D. Gillesen, and V. du Vigneaud, *J. Amer. Chem. Soc.*, **88**, 1310 (1966).

340 mg (59%); mp 238–240°;  $[\alpha]^{20}_D +41.5^\circ$  (*c* 1, DMF); lit.<sup>4</sup> (*L* isomer) mp 239–240°;  $[\alpha]^{20}_D -42^\circ$  (*c* 1, DMF).

**Procedure B.**—Octapeptide·HBr, obtained in a run identical with the one described in procedure A, was dissolved in 60 ml of MeOH and treated with 100 mg of imidazole. The resulting soln was evapd to a solid residue which was dried over P<sub>2</sub>O<sub>5</sub> under vacuum. The dry residue was dissolved in 2 ml of DMF and condensed with *p*-nitrophenyl *S*-benzyl- $\beta$ -mercaptopropionate as described in procedure A: yield 485 mg (84%); mp 241–243°;  $[\alpha]^{19}_D +41.6^\circ$  (*c* 1, DMF). *Anal.* (C<sub>57</sub>H<sub>79</sub>N<sub>11</sub>O<sub>12</sub>S<sub>2</sub>) C, H, N.

**Deamino-D-oxytocin.**—The debenzylation of 200 mg of amide I was performed with Na in 400 ml of liq NH<sub>3</sub> freshly distd from Na.<sup>5</sup> The soln was concd, and the last 30 ml of liq NH<sub>3</sub> was lyophilized. The residual white powder was dissolved in 150 ml of 0.25% AcOH, the pH was adjusted to 6.8 with 1 *N* NH<sub>4</sub>OH, and the resulting clear soln was titrated with 0.011 *M* K<sub>3</sub>Fe(CN)<sub>6</sub> until a yellow color began to appear (27 ml). Then excess ferricyanide (8 ml) was added. After 30 min the soln gave a negative Ellman test, and it was passed through a column of AG3-X4 (Cl<sup>-</sup>). The soln of the crude product thus obtained was divided into 2 equal portions, each of which was purified by a different method. One half of the soln was concd to 15 ml and subjected to countercurrent distribution in the solvent system *n*-BuOH-*n*-PrOH-0.5% AcOH contg 0.1% pyridine (6:1:8). After 200 transfers a main peak (*K* = 4.4) was obtained, as detd by measurement of the Folin-Lowry color values.<sup>14</sup> The contents of tubes 155–170 were combined, concd, and lyophilized to yield 25 mg of a white fluffy powder. A sample was hydrolyzed in 6

*N* HCl at 120° for 20 hr for amino acid analysis,<sup>15</sup> and the following molar ratios were found, with the value of Gly taken as 1.0: Gly, 1.0; Leu, 1.0; Pro, 1.0; Asp, 1.0; Glu, 1.1; Ile, 0.93;  $\alpha$ Ile, 0.07; Tyr, 1.0; Cys, 0.24; the mixed disulfide of cysteine and  $\beta$ -mercaptopropionic acid, 0.67; NH<sub>3</sub>, 2.9.

The other half of the crude soln of deamino-D-oxytocin was concd to a low vol and purified by partition chromatography by the method of Yamashiro.<sup>8</sup> A Sephadex G-25 column (2.15 × 113 cm) was employed with the solvent system *n*-BuOH-C<sub>6</sub>H<sub>6</sub>-3.5% AcOH contg 1.5% pyridine (1:1:2). Elution with the upper phase was performed at a rate of 30 ml/hr. The Folin-Lowry color values showed a main peak with *R*<sub>f</sub> 0.19. The corresp value for deamino-L-oxytocin is 0.19.<sup>14</sup> Fractions corresp to the main peak were combined, concd, and lyophilized: yield 37 mg of white fluffy powder;  $[\alpha]^{20}_D +104^\circ$  (*c* 0.5, 1 *N* AcOH); lit.<sup>4</sup> (amorphous *L* isomer)  $[\alpha]^{21}_D -107^\circ$  (*c* 0.5, 1 *N* AcOH). A sample of deamino-D-oxytocin was hydrolyzed in 6 *N* HCl at 120° for 20 hr for amino acid analysis. The following molar ratios were obtd, with the value of Gly taken as 1.0: Gly, 1.0; Leu, 1.0; Pro, 1.0; Asp, 1.0; Glu, 1.0; Ile, 1.0;  $\alpha$ Ile, 0.02; Tyr, 1.0; Cys, 0.34; the mixed disulfide of cysteine and  $\beta$ -mercaptopropionic acid, 0.62; NH<sub>3</sub>, 2.9. *Anal.* (C<sub>43</sub>-H<sub>65</sub>N<sub>11</sub>O<sub>12</sub>S<sub>2</sub>) C, H, N.

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(14) O. H. Lowry, N. J. Rosebrough, A. L. Farr, and R. J. Randall, *J. Biol. Chem.*, **193**, 265 (1951).

(15) D. H. Spackman, W. H. Stein, and S. Moore, *Anal. Chem.*, **30**, 1190 (1958).

## New Compounds

### $\alpha$ -Bromo- and $\alpha$ -Chloropyridylalanines<sup>1</sup>

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Phenylalanine analogs have exhibited biological activity in certain mammalian phenylalanine, tryptophan, and tyrosine hydroxylase systems.<sup>2–5</sup> *p*-Chlorophenylalanine depletes brain serotonin in the rat<sup>6</sup> thus causing an abnormal psychic behavior of the animal.<sup>7</sup> The synthesis of the  $\alpha$ -fluoro- and  $\alpha$ -hydroxypyridylalanines has been described in an earlier study and certain of these compounds are toxic to the growth of various microorganisms.<sup>8</sup> In this report the synthesis of the bromo and chloro analogs is described.

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(2) J. I. DeGraw, M. Cory, W. A. Skinner, M. C. Theisen, and C. Mitoma, *J. Med. Chem.*, **10**, 64 (1967).

(3) R. W. Fuller, *Life Sci.*, **4**, 1 (1965).

(4) W. S. Saari, J. Williams, S. F. Britcher, D. E. Wolf, and F. A. Kuehl, Jr., *J. Med. Chem.*, **10**, 1008 (1967).

(5) W. F. Coulson, E. Wardle, and J. B. Jepson, *Biochim. Biophys. Acta*, **167**, 99 (1968).

(6) B. K. Koe and A. Weissman, *J. Pharmacol. Exp. Ther.*, **154**, 499 (1966).

(7) A. Tagliamonte, P. Tagliamonte, G. L. Gessa, and B. B. Brodie, *Science*, **166**, 1433 (1969).

(8) P. T. Sullivan, C. B. Sullivan, and S. J. Norton, *J. Med. Chem.*, **14**, 211 (1971).

### Experimental Section

A Thomas-Hoover capillary melting point apparatus was employed for all mp determinations, and the melting points reported are uncorr. Uv spectra were determined with a Beckman DBG recording spectrophotometer. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within  $\pm 0.4\%$  of the theoretical values unless otherwise specified. The aminopicolines were obtained from Aldrich Chemical Co., Inc. and J. T. Baker Laboratory Chemicals.

The following reaction procedures are given for specific compds; compds indicated by reference to the particular table were prepared in like manner.

**$\alpha$ -Bromopicolines.**—The appropriate aminopicoline was diazotized as previously reported<sup>9</sup> utilizing HBr, Br<sub>2</sub>, and NaNO<sub>2</sub>. The boiling points and melting points agreed in all cases with those reported above.

**$\alpha$ -Chloropicolines.**—The appropriate aminopicoline was diazotized as reported<sup>10</sup> employing HCl and NaNO<sub>2</sub>. The boiling points were in agreement with those reported in the literature.

**2-Bromo-3-bromomethylpyridine·HBr (Table I, 1–8).**—2-Bromo-3-methylpyridine (29.2 g, 0.17 mole), NBS (30.2 g, 0.17 mole), and 1.5 g of benzoyl peroxide in 500 ml of MgSO<sub>4</sub>-dried CCl<sub>4</sub> were refluxed several hours. The succinimide was removed by filtration, and the filtrate was concd *in vacuo* to about 100 ml. The soln was washed with 100 ml of each of the following: 4% NaOH, H<sub>2</sub>O, and 2% aq HBr. Et<sub>2</sub>O was added to the org layer to make a total of 175 ml, and the dried soln was satd with anhyd

(9) For syntheses and physical constants of the  $\alpha$ -bromopicolines see F. A. Case, *J. Amer. Chem. Soc.*, **68**, 2574 (1946); P. Adams and S. Miyano, *ibid.*, **76**, 3168 (1954).

(10) For syntheses and physical constants of the  $\alpha$ -chloropicolines cf. O. Seide, *Ber.*, **57B**, 1802 (1924); O. Seide, *ibid.*, **57B**, 791 (1924); O. A. Zeide, *Zh. Russ. Fiz.-Khim. Obschest.*, **50**, 534 (1920); W. Herz and D. R. K. Murty, *J. Org. Chem.*, **26**, 122 (1961).